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Camphor (PIM 095)

Springfield, Illinois, CC Thomas. Aronow R (1976). **Camphor**

Poisoning. J Am Med

Assoc. 235: 126O. ... Arch Toxicol 51: 101-106. Kresel JJ (1982)

Camphor. ...

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The odor of **camphor** on the breath or in urine may assist in diagnosis of overexposure.

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ORS ORT OSE OUD OUR OUT OVA OWE OWL OWN **OXO OXY**
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Thieme-connect / Synthesis / Table of contents

ODA, Masaji; KITAHARA, Yoshio: Reaction of 2-Chlorotropone with Enamines.

Synthesis of 2-(2-Oxocycloalkyl)-tropone PDF (121 kb) ...

www.thieme-connect.com/ejournals/toc/synthesis/1335 - [Similar pages](#)[[More results from www.thieme-connect.com](#)]United States Patent Application: 0050014095

Specific examples of the 2-oxocycloalkyl formed by bonding P.sup.8 and P.sup.9 together with the adjacent -CHCO- include 2-oxocyclohexyl, 2-oxocyclopentyl ...

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... and 5-hydroxytryptamine (5-HT2A) receptors of

1-(benzoylpropyl)-4-(1-oxocycloalkyl-2-ethyl)-piperazines as cyclic butyrophenones derivatives Chem. Phar. ...

www.cajal.csic.es/publicac/publ1996.htm - 59k - [Cached](#) - [Similar pages](#)Patent 4126708: Flavoring with 2-acyl-5-substituted ...

... R.sub.4 and R.sub.8 are the same or different and each represents hydrogen or methyl), hydroxyalkyl, oxoalkyl, hydroxycycloalkyl or oxocycloalkyl having the ...

freepatentsonline.com/4126708.html - 52k - Supplemental Result - [Cached](#) - [Similar pages](#)Patent 5635332: Alkylsulfonium salts and photoresist compositions ...

in R.sup.3, the C.sub.5 -C.sub.7 2-oxocycloalkyl radical includes, ... in particular,

It is believed that the ketone group (2-oxocycloalkyl group) structure ...

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photoactivated 5-FU prodrugs 1-2 were synthesized using A similar method as for

The preparation of 1-(2-oxocycloalkyl)-5-fluorouracils reported Previously. ...

www.rsc.org/.../ArticleLinking.cfm?JournalCode=OB&Year=2005&ManuscriptID=b417734g&Iss=4 -Supplemental Result - [Similar pages](#)Chemical amplifying type positive resist composition and sulfonium ...

... sup.4 represents alkyl or a cycloalkyl, or Q.sup.3 and Q4 form, together with a CHC(O) group to which Q.sup.3 and Q4 are adjacent, a 2-oxocycloalkyl group; and ...

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... 3 to 10 carbon atoms or phenyl optionally substituted by alkyl having 1 to 6 carbon atoms, or Q.sup.3 and Q.sup.4 bond to form 2-oxocycloalkyl together with ...

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Camphor

From Wikipedia, the free encyclopedia

Camphor is a white transparent waxy crystalline solid with a strong penetrating pungent aromatic odor. It is a terpenoid with the chemical formula $C_{10}H_{16}O$. It is found in wood of the camphor laurel (*Cinnamomum camphora*), a large evergreen tree found in Asia (particularly in Borneo, hence its alternate name); it can also be synthetically produced from oil of turpentine. It is used for its scent, as an embalming fluid and for medicinal purposes.

Modern uses include as a plasticizer for cellulose nitrate, as a moth repellent, as an antimicrobial substance, in embalming, and in fireworks. A form of anti-itch gel currently on the market uses camphor as its active ingredient. It is also used in medicine. Camphor is readily absorbed through the skin and produces a feeling of cooling similar to that of menthol and acts as slight local anesthetic and antimicrobial substance. It may also be administered orally in small quantities (50 mg) for minor heart symptoms and fatigue. Camphor is also used as a flavoring in sweets in India and Europe. It is thought that camphor was used as a flavouring in confections resembling ice cream in China during the Tang dynasty (A.D. 618-907).

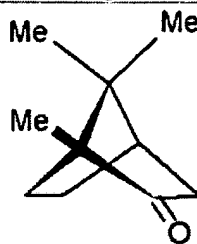
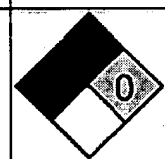
In larger quantities, it is poisonous when ingested and can cause seizures, confusion, irritability, and neuromuscular hyperactivity. In 1980, the United States Food and Drug Administration set a limit of 11% allowable camphor in consumer products and totally banned products labeled as camphorated oil, camphor oil, camphor liniment, and camphorated liniment (but camphor is absent in "white camphor essential oil"). Since alternative treatments exist, medicinal use of camphor is discouraged by the FDA, except for skin-related uses, such as medicated powders, which contain only small amounts of camphor.

Other substances deriving from trees are sometimes wrongly sold as camphor.

The word camphor derives from the Malay word *kapur*, meaning "camphor tree" [1]

(<http://www.etymonline.com/index.php?search=camphor&searchmode=none>).

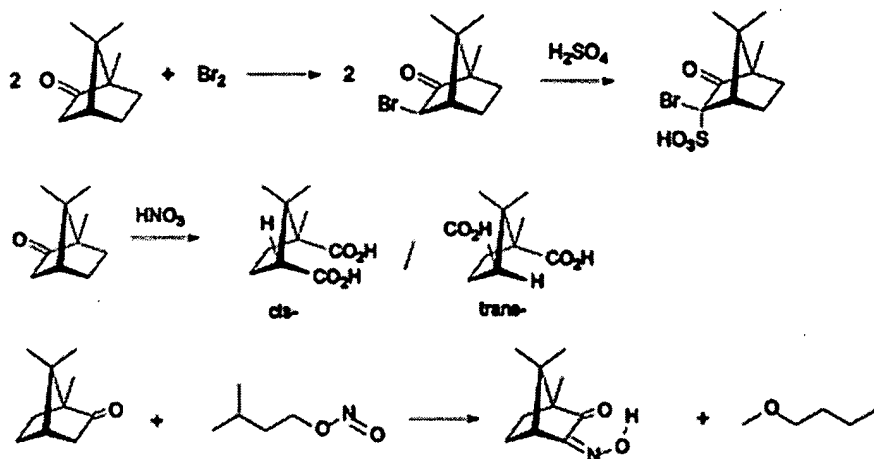
Camphor was the first synthesized by Gustaf Komppa in

Camphor	
	
General	
Systematic name	1,7,7-trimethylbicyclo[2.2.1]heptan-2-one
Other names	2-bornanone, 2-camphanone, bornan-2-one, Formosa
Molecular formula	$C_{10}H_{14}O$
SMILES	<chem>CC1(C)C2(C)C(CC1CC2)=O</chem>
Molar mass	152.23 g/mol
Appearance	White or colourless crystals
CAS number	[76-22-2] (unspecified) [464-49-3] ((1 <i>R</i>)-Camphor) [464-48-2] ((1 <i>S</i>)-Camphor)
Properties	
Density and phase	0.990, solid
Solubility in water	0.12 g in 100 ml
Solubility in acetic acid	~200 g in 100 ml
Solubility in ethanol	~100 g in 100 ml
Solubility in acetone	~250 g in 100 ml
Solubility in ether	~100 g in 100 ml
Solubility in chloroform	~200 g in 100 ml
Melting point	179.75 °C (452.9 K)
Boiling point	204 °C (477 K)
Acidity (pK_a)	?
Chiral rotation $[\alpha]_D$	+44.1°, (1 <i>R</i>)-Camphor
Hazards	
MSDS	External MSDS
Main hazards	flammable
NFPA 704	
R-phrases	11-20/21/22-36/37/38
S-phrases	16-26-36
RTECS number	EX1260000 (<i>R</i>) EX1250000 (<i>S</i>)
Supplementary data page	
Structure and	n , ϵ , etc.

1903. Previously, some organic compounds (such as urea) had been synthesized in the laboratory as a proof of concept, but camphor was a scarce natural product with a worldwide demand. The synthesis was the first industrial total synthesis, when Komppa began industrial production in Tainionkoski, Finland in 1907.

Reactions

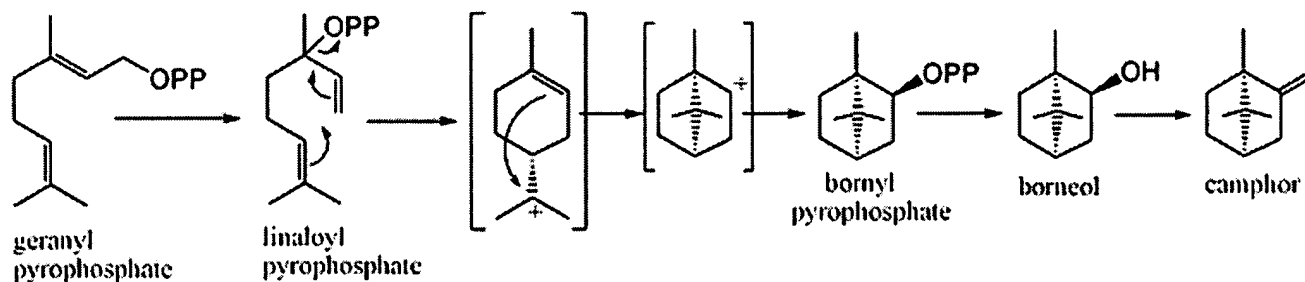
properties	
Thermodynamic data	Phase behaviour Solid, liquid, gas
Spectral data	UV, IR, NMR, MS
Related compounds	
Related ketones	fenchone, thujone
Related compounds	camphene, pinene borneol, isoborneol 10-Camphorsulfonic acid
Except where noted otherwise, data are given for materials in their standard state (at 25 °C, 100 kPa) Infobox disclaimer and references	



Camphor can also be reduced to isoborneol using sodium borohydride.

Biosynthesis

Camphor is produced from geranyl pyrophosphate, via cyclisation of linaloyl pyrophosphate to bornyl pyrophosphate, followed by hydrolysis to borneol and oxidation to camphor.



References

1. J. Mann, R. S. Davidson, J. B. Hobbs, D. V. Banthorpe, J. B. Harborne, *Natural Products*, pp. 309-311, Addison Wesley Longman Ltd., Harlow, UK, 1994. ISBN 0582060095.
2. *Handbook of Chemistry and Physics*, CRC Press, Ann Arbor, Michigan.
3. *The Merck Index*, 7th edition, Merck & Co, Rahway, New Jersey, USA, 1960.

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Categories: Pyrotechnic chemicals | Ketones | Terpenes and terpenoids | Materials involved in Hinduism

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S3	2561	S1 and (\$sulfonium\$ \$sulphonium\$) cancel	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/01 19:57
S1 was same as S6 below but disappeared upon refresh and was cancelled as unneeded so removed from Hist. List unententionally						
S4	27846	(\$sulfonium\$ \$sulphonium\$)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/01 16:59
S5	4411	(\$sulfonium\$ \$sulphonium\$)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/01 16:57
S6	5043	(430/270.1).CCLS.	US-PGPUB; USPAT	OR	OFF	2006/02/01 16:58
S7	16	(\$sulfonium\$ \$sulphonium\$) looked at all ^{2/5/06}	IBM_TDB	OR	ON	2006/02/01 17:02
S8	2918	(\$sulfonium\$ \$sulphonium\$)	DERWENT	OR	ON	2006/02/01 17:02
S9	542	(\$sulfonium\$ \$sulphonium\$) and g03f\$.ipc.	DERWENT	OR	ON	2006/02/01 17:02
S10	1	1997-179149.NRAN.	DERWENT	OR	OFF	2006/02/01 18:23
S11	2561	S1 and (\$sulfonium\$ \$sulphonium\$) not S9	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/02 15:51
S12	5049	(430/270.1).CCLS.	US-PGPUB; USPAT	OR	OFF	2006/02/02 15:52
S13	542	(\$sulfonium\$ \$sulphonium\$) and g03f\$.ipc.	DERWENT	OR	ON	2006/02/02 15:52
S14	2278	S12 and (\$sulfonium\$ \$sulphonium\$) AND (positiv\$5 chemic\$ near5 amplif\$10) not S13	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/04 20:06
S15	2278	(S12 and (\$sulfonium\$ \$sulphonium\$) AND (positiv\$5 chemic\$ near5 amplif\$10)) not S13	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/02 15:57
S16	510	(430/921.922,914).CCLS.	US-PGPUB; USPAT	OR	OFF	2006/02/04 14:04
S17	412	S16 AND (\$SULFONIUM\$ \$SULPHONIUM\$)	US-PGPUB; USPAT	OR	OFF	2006/02/04 14:05
S18	1517	"522"/\$.CCLS. AND (\$SULFONIUM\$ \$SULPHONIUM\$)	US-PGPUB; USPAT	OR	OFF	2006/02/04 14:06
S19	1825	S17 OR S18	US-PGPUB; USPAT	OR	OFF	2006/02/04 14:06
S20	510	(430/921.922,914).CCLS:	US-PGPUB; USPAT	OR	OFF	2006/02/04 19:49
S21	412	S20 AND (\$SULFONIUM\$ \$SULPHONIUM\$)	US-PGPUB; USPAT	OR	OFF	2006/02/04 19:49

S22	1517	"522"\$.CCLS. AND (\$SULFONIUM\$ \$SULPHONIUM\$)	US-PGPUB; USPAT	OR	OFF	2006/02/04 19:49
S23	1825	S21 OR S22	US-PGPUB; USPAT	OR	OFF	2006/02/04 19:49
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S30	CH 71	430/280.1.CCLS. and (\$sulfonium\$ \$sulphonium\$) AND (positiv\$5 chemic\$ near5 amplif\$10) not S27 NOT 430/270.1,281.1. CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/04 20:23
S31	CH 15	430/285.1.CCLS. and (\$sulfonium\$ \$sulphonium\$) AND (positiv\$5 chemic\$ near5 amplif\$10) not S27 NOT 430/270.1,281.1,280. 1.CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/02/04 20:23

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